i-STAT CG8+ Cartridge

Intended for use with the i-STAT Alinity Instrument

NAME

i-STAT CG8+ Cartridge - REF 03P88-25

© SIAT CG8+

INTENDED USE

The i-STAT CG8+ cartridge with the i-STAT Alinity System is intended for use in the *in vitro* quantification of sodium, potassium, ionized Calcium, glucose, hematocrit, pH, partial pressure of oxygen, and partial pressure of carbon dioxide in arterial, venous or capillary whole blood.

Analyte	Intended Use			
Sodium (Na)	Sodium measurements are used for monitoring electrolyte imbalances.			
Potassium (K)	Potassium measurements are used in the diagnosis and monitoring of diseases and clinical conditions that manifest high and low potassium levels.			
Ionized Calcium (iCa)	lonized calcium measurements are used in the diagnosis, monitoring, and treatment of conditions including, but not limited to, parathyroid disease, a variety of bone diseases, chronic renal disease, tetany, and disturbances related to surgical and intensive care.			
Glucose (Glu)	Glucose measurements are used in the diagnosis, monitoring, and treatment of carbohydrate metabolism disorders including, but not limited to, diabetes mellitus, neonatal hypoglycemia, idiopathic hypoglycemia, and pancreatic islet cell carcinoma.			
Hematocrit (Hct)	Hematocrit measurements can aid in the determination and monitoring of normal or abnormal total red cell volume status including, but not limited to, conditions such as anemia, erythrocytosis, and blood loss related to trauma and surgery.			
pH Partial Pressure of Oxygen (PO2)	pH, P O ₂ , and P CO ₂ measurements are used in the diagnosis, monitoring, and treatment of respiratory disturbances and metabolic and respiratory-based acid-base disturbances.			
Partial Pressure of Carbon Dioxide (P CO ₂)	Bicarbonate is used in the diagnosis and treatment of numerous potentially serious disorders associated with changes in body acid-base balance.			

SUMMARY AND EXPLANATION / CLINICAL SIGNIFICANCE

Measured:

Sodium (Na)

Tests for sodium in the blood are important in the diagnosis and treatment of patients suffering from hypertension, renal failure or impairment, cardiac distress, disorientation, dehydration, nausea and diarrhea. Some causes of increased values for sodium include dehydration, diabetes insipidus, salt poisoning, skin losses, hyperaldosteronism and CNS disorders. Some causes for decreased values for sodium include dilutional hyponatremia (cirrhosis), depletional hyponatremia and syndrome of inappropriate ADH.

Potassium (K)

Tests for potassium in the blood are important in the diagnosis and treatment of patients suffering from hypertension, renal failure or impairment, cardiac distress, disorientation, dehydration, nausea and diarrhea. Some causes of increased values for potassium include renal glomerular disease, adrenocortical insufficiency, diabetic ketoacidosis (DKA), sepsis and *in vitro* hemolysis. Some causes of decreased values for potassium include renal tubular disease, hyperaldosteronism, treatment of DKA, hyperinsulinism, metabolic alkalosis and diuretic therapy.

Ionized Calcium (iCa)

Although most of the calcium in blood is bound to protein or complexed to smaller anionic species, the biologically active fraction of calcium is free ionized calcium. Through its role in a number of enzymatic reactions and in membrane transport mechanisms, ionized calcium is vitally important in blood coagulation, nerve conduction, neuromuscular transmission and in muscle contraction. Increased ionized calcium (hypercalcemia) may result in coma. Other symptoms reflect neuromuscular disturbances, such as hyperreflexia and/or neurologic abnormalities such as neurasthenia, depression or psychosis. Decreased ionized calcium (hypocalcemia) often results in cramps (tetany), reduced cardiac stroke work and depressed left ventricular function. Prolonged hypocalcemia may result in bone demineralization (osteoporosis) which can lead to spontaneous fractures. Measurements of ionized calcium have proven of value under the following clinical conditions: transfusion of citrated blood, liver transplantation, open heart surgery, neonatal hypocalcemia, renal disease, hyperparathyroidism, malignancy, hypertension and pancreatitis.

Glucose (Glu)

Glucose is a primary energy source for the body and the only source of nutrients for brain tissue. Measurements for determination of blood glucose levels are important in the diagnosis and treatment of patients suffering from diabetes and hypoglycemia. Some causes for increased values of glucose include diabetes mellitus, pancreatitis, endocrine disorders (e.g., Cushing's syndrome), drugs (e.g., steroids, thyrotoxicosis), chronic renal failure, stress, or I.V. glucose infusion. Some causes of decreased values of glucose include insulinoma, adrenocortical insufficiency, hypopituitarism, massive liver disease, ethanol ingestion, reactive hypoglycemia, and glycogen storage disease.

Hematocrit (Hct)

Hematocrit is a measurement of the fractional volume of red blood cells. This is a key indicator of the body's state of hydration, anemia or severe blood loss, as well as the blood's ability to transport oxygen. A decreased hematocrit can be due to either overhydration, which increases the plasma volume, or a decrease in the number of red blood cells caused by anemias or blood loss. An increased hematocrit can be due to loss of fluids, such as in dehydration, diuretic therapy, and burns, or an increase in red blood cells, such as in cardiovascular and renal disorders, polycythemia vera, and impaired ventilation.

Hq

pH is an index of the acidity or alkalinity of the blood with an arterial pH of <7.35 indicating an acidemia and >7.45 alkalemia. ¹

Partial Pressure of Oxygen (PO₂)

 PO_2 (partial pressure of oxygen) is a measurement of the tension or pressure of oxygen dissolved in blood. Some causes for decreased values of PO_2 include decreased pulmonary ventilation (e.g., airway obstruction or trauma to the brain), impaired gas exchange between alveolar air and pulmonary capillary blood (e.g., bronchitis, emphysema, or pulmonary edema), and alteration in the flow of blood within the heart or lungs (e.g., congenital defects in the heart or shunting of venous blood into the arterial system without oxygenation in the lungs).

Partial Pressure of Carbon Dioxide (PCO₂)

 PCO_2 along with pH is used to assess acid-base balance. PCO_2 (partial pressure of carbon dioxide), the respiratory component of acid-base balance, is a measure of the tension or pressure of carbon dioxide dissolved in the blood. PCO_2 represents the balance between cellular production of CO_2 and ventilatory removal of CO_2 and a change in PCO_2 indicates an alteration in this balance. Causes of primary respiratory acidosis (increase in PCO_2) are airway obstruction, sedatives and anesthetics, respiratory distress syndrome, and chronic obstructive pulmonary disease. Causes of primary respiratory alkalosis (decreased PCO_2) are hypoxia (resulting in hyperventilation) due to chronic heart failure, edema and neurologic disorders, and mechanical hyperventilation.

TEST PRINCIPLE

The i-STAT System uses direct (undiluted) electrochemical methods. Values obtained by direct methods may differ from those obtained by indirect (diluted) methods. ²

Measured:

Sodium (Na), Potassium (K) and ionized Calcium (iCa)

The respective analyte is measured by ion-selective electrode potentiometry. In the calculation of results, concentration is related to potential through the Nernst equation.

Glucose (Glu)

Glucose is measured amperometrically. Oxidation of glucose, catalyzed by the enzyme glucose oxidase, produces hydrogen peroxide (H_2O_2). The liberated H_2O_2 is oxidized at the electrode to produce a current proportional to the sample glucose concentration.

$$β$$
-D-glucose + $H_2O + O_2$ D-gluconic acid + H_2O_2

$$H_2O_2$$
 $2H^+ + O_2 + 2e^-$

Hematocrit (Hct)

Hematocrit is determined conductometrically. The measured conductivity, after correction for electrolyte concentration, is inversely related to the hematocrit.

На

pH is measured by direct potentiometry. In the calculation of results for pH, concentration is related to potential through the Nernst equation.

PO_2

 PO_2 is measured amperometrically. The oxygen sensor is similar to a conventional Clark electrode. Oxygen permeates through a gas permeable membrane from the blood sample into an internal electrolyte solution where it is reduced at the cathode. The oxygen reduction current is proportional to the dissolved oxygen concentration.

PCO₂

 PCO_2 is measured by direct potentiometry. In the calculation of results for PCO_2 , concentration is related to potential through the Nernst equation.

Temperature "Correction" Algorithm

pH, $\dot{P}O_2$, and $\dot{P}CO_2$ are temperature-dependent quantities and are measured at 37°C. The pH, $\dot{P}O_2$, and $\dot{P}CO_2$ readings at a body temperature other than 37°C can be 'corrected' by entering the patient's temperature on the chart page of the analyzer. In this case, blood gas results will be displayed at both 37°C and the patient's temperature.

The pH, PO_2 , and PCO_2 at the patient's temperature (T_p) are calculated as follows 3 :

$$pH(T_p) = pH - 0.0147(T_p - 37) + 0.0065(7.4 - pH)(T_p - 37)$$

$$PO_{2}(T_{p}) = PO_{2} \times 10^{\frac{5.49 \times 10^{-11} PO_{2}^{3.88} + 0.071}{9.72 \times 10^{-9} PO_{2}^{3.88} + 2.30}} (T_{p} - 37)$$

$$PCO_2(T_p) = PCO_2 \times 10^{0.019(T_p - 37)}$$

Calculated:

HCO₃, TCO₂, and BE

- HCO₃ (bicarbonate), the most abundant buffer in the blood plasma, is an indicator of the buffering capacity of blood. Regulated primarily by the kidneys, HCO₃ is the metabolic component of acid-base balance.
- TCO₂ is a measure of carbon dioxide which exists in several states: CO₂ in physical solution or loosely bound to proteins, bicarbonate (HCO₃) or carbonate (CO₃) anions, and carbonic acid (H₂CO₃). Measurement of TCO₂ as part of an electrolyte profile is useful chiefly to evaluate HCO₃ concentration. TCO₂ and HCO₃ are useful in the assessment of acid-base imbalance (along with pH and PCO₂) and electrolyte imbalance.
- The calculated TCO₂ provided by the i-STAT System is determined from the measured and reported values of pH and PCO₂ according to a simplified and standardized form of the Henderson-Hasselbalch equation.³
- This calculated TCO₂ measurement is metrologically traceable to the i-STAT pH and PCO₂ measurements, which are in turn traceable to primary standard reference materials for pH and PCO₂. Like all calculated parameters reported by the i-STAT System, the user can independently determine TCO₂ values from the reported pH and PCO₂ measurements using a combination of the equation for HCO₃ given in the PCO₂.
- Base excess of the extracellular fluid (ECF) or standard base excess is defined as the concentration of titratable base minus the concentration of titratable acid when titrating the average ECF (plasma plus interstitial fluid) to an arterial plasma pH of 7.40 at PCO₂ of 40 mmHg at 37 °C. Excess concentration of base in the average ECF remains virtually constant during acute changes in the PCO₂ and reflects only the non-respiratory component of pH-disturbances.

When a cartridge includes sensors for both pH and PCO_2 , bicarbonate (HCO₃), total carbon dioxide (TCO₂) and base excess (BE) are calculated.³

 $log HCO_3 = pH + log PCO_2 - 7.608$

```
TCO_2 = HCO_3 + 0.03 PCO_2 BE_{ecf} = HCO_3-24.8 + 16.2(pH-7.4) BE_b = (1 - 0.014*Hb) * [HCO_3 - 24.8 + (1.43 * Hb + 7.7) * (pH - 7.4)]
```

SO_2

- sO₂ (oxygen saturation) is the amount of oxyhemoglobin expressed as a fraction of the total amount of hemoglobin able to bind oxygen (oxyhemoglobin plus deoxyhemoglobin).
- sO₂ is calculated from measured **P**O₂ and pH and from HCO₃ calculated from measured **P**CO₂ and pH. However, this calculation assumes normal affinity of oxygen for hemoglobin. It does not take into account erythrocyte diphosphoglycerate (2,3-DPG) concentrations which affect the oxygen dissociation curve. The calculation also does not take into account the effects of fetal hemoglobin or dysfunctional hemoglobins (carboxy-, met-, and sulfhemoglobin). Clinically significant errors can result from incorporation of such an estimated sO₂ value for oxygen saturation in further calculations, such as shunt fraction, or by assuming the value obtained is equivalent to fractional oxyhemoglobin.

$$SO_2 = 100$$
 $\frac{(X^3 + 150X)}{X^3 + 150X + 23400}$
where $X = PO_3 \cdot 10^{(0.48(pH-7.4)-0.0013(HCO_3-25))}$

Hemoglobin

The i-STAT System provides a calculated hemoglobin result which is determined as follows 4:

hemoglobin (g/dL) = hematocrit (% PCV) x 0.34

hemoglobin (g/dL) = hematocrit (decimal fraction) x 34

To convert a hemoglobin result from g/dL to mmol/L, multiply the displayed result by 0.621. The calculation of hemoglobin from hematocrit assumes a normal MCHC.

See below for information on factors affecting results. Certain substances, such as drugs, may affect analyte levels in vivo. ⁵ If results appear inconsistent with the clinical assessment, the patient sample should be retested using another cartridge.

REAGENTS

Contents

Each i-STAT cartridge contains one reference electrode, sensors for the measurement of specific analytes, and a buffered aqueous calibrant solution that contains known concentrations of analytes and preservatives. A list of reactive ingredients relevant for the i-STAT CG8+ cartridge is indicated below:

Sensor Reactive Ingredient		Biological Source	Minimum Quantity
Na	Na Sodium (Na+)		121 mmol/L
К	Potassium (K+)	N/A	3.6 mmol/L
iCa	Calcium (Ca ²⁺)	N/A	0.9 mmol/L
Glu	Glucose	N/A	7 mmol/L
Giu	Glucose Oxidase	Aspergillus niger	0.002 IU
pH Hydrogen Ion (H ⁺)		N/A	6.66 pH
PCO ₂ Carbon Dioxide (CO ₂)		N/A	25.2 mmHg

Warnings and Precautions

- For in vitro diagnostic use.
- Cartridges are intended for single-use only. Do not reuse.
- Refer to the i-STAT Alinity System Operations Manual for all warnings and precautions.

Storage Conditions

- Refrigeration at 2–8 °C (35–46 °F) until expiration date.
- Room Temperature at 18–30 °C (64–86 °F). Refer to the cartridge box for room temperature storage requirements.

INSTRUMENTS

The i-STAT CG8+ cartridge is intended for use with the i-STAT Alinity Instrument (Model No. AN-500).

SPECIMEN COLLECTION AND PREPARATION FOR ANALYSIS

Specimen Types

Arterial, venous or capillary whole blood.

Sample Volume: 95 µL

Blood Collection Options and Test Timing (time from collection to cartridge fill)
As higher heparin-to-blood ratios may affect results, fill blood collection tubes and syringes to capacity, following manufacturers' instructions.

g manufacturers' instructions.
CG8+ Sample Collection
Without anticoagulant
Maintain anaerobic conditions prior to filling this cartridge.
Mix sample immediately before filling cartridge.
Fill cartridge within 3 minutes of sample collection.
With balanced heparin anticoagulant
Maintain anaerobic conditions prior to filling this cartridge.
Mix sample immediately before filling cartridge.
Fill cartridge within 10 minutes of sample collection.
Without anticoagulant
Maintain anaerobic conditions prior to filling this cartridge.
Mix sample immediately before filling cartridge.
Fill cartridge within 3 minutes of sample collection.
·
With lithium heparin anticoagulant
Maintain anaerobic conditions prior to filling this cartridge.
Mix sample immediately before filling cartridge.
Fill cartridge within 10 minutes of sample collection.
With balanced heparin anticoagulant
Mix sample immediately before filling cartridge.
Fill cartridge within 3 minutes of sample collection.
Not Recommended

PROCEDURE FOR CARTRIDGE TESTING

Preparation for Use:

- 1. Individual cartridges may be used after standing five minutes at room temperature. An entire box of cartridges should stand at room temperature for one hour.
- 2. All cartridges should be used immediately after opening pouch.
- 3. If the pouch has been punctured, the cartridge should not be used.
- 4. Do not return cartridges to the refrigerator after bringing them to room temperature.

How to Perform Patient Testing

- 1. From the Home screen, touch "Perform Patient Test". This initiates the patient testing pathway.
- 2. To begin, follow instructions on the screen to "Scan or Enter OPERATOR ID"
- 3. Follow instructions on the screen to "Scan or Enter PATIENT ID"
- 4. Continue to follow prompts on the screen to proceed with patient testing. "Scan (CARTRIDGE POUCH)

 Barcode", Scanning is required. Information cannot be entered manually.
- 5. The screen for selecting sample type will display if more than one sample type is applicable; select sample type if applicable.
- 6. Follow instructions on the screen to "Close and Insert Filled Cartridge". The action buttons at the bottom of the screen allow forward, backward and pause functionality.
- 7. Once the cartridge is inserted, "Contacting Cartridge" will display followed by the countdown bar. The following alerts are also displayed: "Cartridge locked in instrument. Do not attempt to remove the Cartridge" and "Testing Instrument Must Remain Level".
- 8. When the test is complete, the test results are displayed.

Analysis Time

Approximately 130-200 seconds.

Quality Control

The i-STAT Alinity System quality control regimen comprises various aspects, with a system design that reduces the opportunity for error, including:

- 1. The i-STAT Alinity System automatically runs a comprehensive set of quality checks of analyzer and cartridge performance each time a sample is tested. This internal quality system will suppress results if the analyzer or cartridge does not meet certain internal specifications.
- **2.** Aqueous-based control solutions are available for verifying the integrity of newly received cartridges.
- 3. In addition, the instrument performs internal electronic checks and calibration during each test cycle, and the Electronic Simulator test provides an independent check on the ability of the instrument to take accurate and sensitive measurements of voltage, current and resistance from the cartridge. The instrument will pass or fail this electronic test depending on whether or not it measures these signals within limits specified in the instrument software.

For additional information on Quality Control, refer to the i-STAT Alinity System Operations Manual located at www.pointofcare.abbott.

Calibration Verification

Standardization is the process by which a manufacturer establishes "true" values for representative samples. A multi-point calibration is derived for each sensor by this standardization process. These calibration curves are stable over many lots.

A one-point calibration is performed each time a cartridge requiring calibration is used. During the first part of the testing cycle, the calibrant solution is automatically released from its foil pack and is positioned over

the sensors. The signals produced by the sensors' responses to the calibrant solution are measured. This one-point calibration adjusts the offset of the stored calibration curve. Next, the instrument automatically moves the sample over the sensors and the signals produced by the sensors' responses to the sample are measured. While coefficients are used rather than graphic calibration curves, the calculation of the result is equivalent to reading the sample's concentration from an adjusted calibration curve.

EXPECTED VALUES

EXI LOTED VAL			BEEEBEN	0E BANGE	
		REPORTABLE	REFERENCE RANGE		
TEST	UNITS *	RANGE	arterial	venous	
MEASURED					
Na	mmol/L(mEq/L)	100-180	138-146 ⁶		
K	mmol/L(mEq/L)	2.0-9.0	3.5–4	1.9 ⁶ **	
iCo	mmol/L	0.25-2.50	1.12-	-1.32 ⁷	
iCa	mg/dL	1.0-10.0	4.5-	-5.3 ⁷	
	mmol/L	1.1–38.9	3.9-	-5.8 ⁷	
Glu	mg/dL	20-700	70-	105 ⁷	
	g/L	0.20-7.00	0.70-	1.05 7	
Homotoprit/Hot	% PCV ***	15–75	38-5	1 6 ****	
Hematocrit/Hct	Fraction	0.15-0.75	0.38-	-0.51 ⁶	
рН		6.50 - 8.20	7.35 - 7.45 ⁷	7.31 - 7.41*****	
P O ₂	mmHg	5 - 800	80 - 105 ⁶ *****		
	kPa	0.7 - 106.6	10.7 - 14.0 ⁶ *****		
P CO ₂	mmHg	5 - 130	35 - 45 ⁷	41 - 51	
	kPa	0.67 - 17.33	4.67 - 6.00	5.47 - 6.80	
CALCULATED					
	g/dL	5.1-25.5	12–1	7 6****	
Hemoglobin/Hb	g/L	51–255	120-	-170 ⁶	
	mmol/L	3.2-15.8	7–	11 ⁶	
Bicarbonate/	mmol/L	1.0 – 85.0	22 – 26*****	23 – 28*****	
HCO₃	(mEq/L)	1.0 - 65.0	22 – 20	25 – 20	
TCO ₂			24 - 29		
	(mEq/L)	0 - 30		27 20	
Base Excess/ BE	mmol/L (mEq/L)	(-30) – (+30)	$(-2) - (+3)^{7}$ $(-2) - (+3)^{7}$		
sO ₂	%	0-100	95 – 98		

^{*} The i-STAT System can be configured with the preferred units. Not applicable for pH test.

Unit Conversion:

- **Ionized Calcium (iCa):** To convert mmol/L to mg/dL, multiply the mmol/L value by 4. To convert mmol/L to mEq/L, multiply the mmol/L value by 2.
- Glucose (Glu): To convert mg/dL to mmol/L, multiply the mg/dL value by 0.055.
- **Hematocrit (Hct):** To convert a result from % PCV (packed cell volume) to fraction packed cell volume, divide the % PCV result by 100. For the measurement of hematocrit, the i-STAT System can be customized to agree with methods calibrated by the microhematocrit reference method using either K₃EDTA or K₂EDTA anticoagulant. Mean cell volumes of K₃EDTA anticoagulated blood are approximately 2–4% less than K₂EDTA anticoagulated blood. While the choice of

^{**} The reference range for potassium has been reduced by 0.2 mmol/L from the range cited in Reference 6 to account for the difference in results between serum and plasma results.

^{***} PCV, packed cell volume.

^{****} The reference ranges for hematocrit and hemoglobin span both female and male populations

^{*****} The reference ranges shown are for a healthy population. Interpretation of blood gas measurements depend on the underlying condition (e.g., patient temperature, ventilation, posture and circulatory status).

****** Calculated from Siggard-Andersen nomogram. 1

anticoagulant affects the microhematocrit method to which all hematocrit methods are calibrated, results from routine samples on hematology analyzers are independent of the anticoagulant used. Since most clinical hematology analyzers are calibrated by the microhematocrit method using K₃EDTA anticoagulant, the i-STAT System default customization is K₃EDTA.

 PO₂ and PCO₂: To convert PO₂ and PCO₂ results from mmHg to kPa, multiple the mmHg value by 0.133.

i-STAT Alinity does not have default reference ranges programmed into the instrument. The reference ranges shown above are intended to be used as guides for the interpretation of results. Since reference ranges may vary with demographic factors such as age, gender and heritage, it is recommended that reference ranges be determined for the population being tested.

METROLOGICAL TRACEABILITY

The measured analytes in the i-STAT CG8+ cartridge are traceable to the following reference materials or methods. The i-STAT System controls and calibration verification materials are validated for use only with the i-STAT System and assigned values may not be commutable with other methods.

Sodium (Na) and Potassium (K) and Ionized Calcium (iCa)

The respective analyte values assigned to i-STAT System controls and calibration verification materials are traceable to the U.S. National Institute of Standards and Technology (NIST) standard reference material SRM956.

Glucose (Glu)

The i-STAT System test for glucose measures glucose amount-of-substance concentration in the plasma fraction of arterial, venous, or capillary whole blood (dimension mmol L⁻¹) for *in vitro* diagnostic use. Glucose values assigned to i-STAT System controls and calibration verification materials are traceable to the U.S. National Institute of Standards and Technology (NIST) standard reference material SRM965.

Hematocrit (Hct)

The i-STAT System test for hematocrit measures packed red blood cell volume fraction in arterial, venous, or capillary whole blood (expressed as the % packed cell volume) for in vitro diagnostic use. Hematocrit values assigned to i-STAT System working calibrators are traceable to the Clinical and Laboratory Standards Institute (CLSI) H7-A3 procedure for determining packed cell volume by the microhematocrit method. 8

pН

The i-STAT System test for pH measures the hydrogen ion amount-of-substance concentration in the plasma fraction of arterial, venous, or capillary whole blood (expressed as the negative logarithm of the relative molal hydrogen ion activity) for *in vitro* diagnostic use. pH values assigned to i-STAT System controls and calibration verification materials are traceable to the U.S. National Institute of Standards and Technology (NIST) standard reference materials SRMs 186-I, 186-II, 185, and 187.

PO_2

The i-STAT System test for partial pressure of oxygen measures the partial pressure of oxygen in arterial, venous, or capillary whole blood (dimension kPa) for *in vitro* diagnostic use. **PO**₂ values assigned to i-STAT System controls and calibration verification materials are traceable to U.S. National Institute of Standards and Technology (NIST) standard reference materials via commercially available certified specialty medical gas standards.

PCO₂

The i-STAT System test for partial pressure of carbon dioxide measures the partial pressure of carbon dioxide in arterial, venous, or capillary whole blood (dimension kPa) for *in vitro* diagnostic use. **P**CO₂ values assigned to i-STAT System controls and calibration verification materials are traceable to U.S. National Institute of Standards and Technology (NIST) standard reference materials via commercially available certified specialty medical gas standards.

Additional information regarding metrological traceability is available from Abbott Point of Care Inc.

PERFORMANCE CHARACTERISTICS

The performance data summarized for Sodium, Glucose and Hematocrit was collected by professionals trained in the use of the i-STAT Alinity System and comparative methods. The performance data summarized for all other tests listed below was collected at Abbott Point of Care. Representative cartridges were used to collect the data.

Precision*

A multiday precision study was performed with aqueous calibration verification materials in representative

cartridges. Duplicates of each aqueous fluid were tested twice a day for 20 days.

Test	Units	Aqueous Cal Ver	n	e a day for Mean	SD (Standard Deviation)	CV (%) [Coefficient of Variation (%)]
Na	mmol/L	Very Low Abnormal	80	99.5	0.32	0.3
	or	Low Abnormal	80	121.2	0.32	0.3
	mEq/L	Normal	80	133.7	0.34	0.3
		High Abnormal	80	160.8	0.38	0.2
		Very High Abnormal	80	180.2	0.56	0.3
K	mmol/L	Very Low Abnormal	80	2.31	0.010	0.4
		Low Abnormal	80	2.90	0.015	0.5
		Normal	80	3.81	0.023	0.6
		High Abnormal	80	6.16	0.026	0.4
		Very High Abnormal	80	7.81	0.039	0.5
iCa	mmol/L	Very Low Abnormal	80	0.32	0.006	2.0
		Low Abnormal	80	0.82	0.008	1.0
		Normal	80	1.29	0.012	1.0
		High Abnormal	80	1.56	0.015	1.0
		Very High Abnormal	80	2.38	0.027	1.1
Glu	mg/dL	Very Low Abnormal	80	26.9	0.42	1.6
		Low Abnormal	80	41.0	0.34	0.8
		High Abnormal	80	125.0	0.32	0.3
		Very High Abnormal	80	286.7	0.77	0.3
		Highest Abnormal	80	600.6	3.47	0.6
Hct	%PCV	Very Low Abnormal	80	16.9	0.46	2.7
		Low Abnormal	80	33.9	0.51	1.5
		High Abnormal	80	55.2	0.49	0.9
		Very High Abnormal	80	65.0	0.39	0.6
рН		Very Low Abnormal	80	6.562	0.005	0.08
		Low Abnormal	80	7.031	0.004	0.06
		Normal	80	7.469	0.003	0.04
		High Abnormal	80	7.769	0.003	0.04
		Very High Abnormal	80	7.986	0.004	0.05
P O ₂	mmHg	Very Low Abnormal	80	72.1	2.02	2.80
		Low Abnormal	80	84.2	1.60	1.90
		Normal	80	118.8	2.10	1.77
		High Abnormal	80	152.1	3.49	2.29

Test	Units	Aqueous Cal Ver	n	Mean	SD (Standard Deviation)	CV (%) [Coefficient of Variation (%)]
		Very High Abnormal	80	377.1	8.52	2.26
P CO ₂	mmHg	Very Low Abnormal	80	17.4	0.43	2.5
		Low Abnormal	80	21.7	0.40	1.8
		Normal	80	28.7	0.57	2.0
		High Abnormal	80	56.2	1.18	2.1
		Very High Abnormal	80	84.5	1.93	2.3

^{*}Note: Representative data, individual laboratories may vary from these data.

Method Comparison

Method comparison was demonstrated in a study comparing the i-STAT Alinity to the i-STAT 1 Wireless (i-STAT 1W) using representative cartridges. The studies were based on CLSI guideline EP9-A3.

Whole blood samples anticoagulated with lithium heparin were evaluated. Samples were analyzed in duplicate on both systems. A weighted Deming regression analysis was performed using the first replicate result from the i-STAT Alinity versus the mean of the duplicates from the i-STAT 1W.

In the method comparison table, n is the number of specimens, and r is the correlation coefficient.

Test	Units		Comparative Method i-STAT 1W
Na	mmol/L	n	174
		Slope	1.0
		r	0.999
		intercept	-1
		X_{min}	115
		X _{max}	173
K	mmol/L	n	195
		Slope	1.00
		r	1.00
		intercept	-0.01
		X_{min}	2.0
		X _{max}	9.0
	mmol/L	n	194
iCa		Slope	1.005
		r	1.000
		intercept	-0.001
		X_{min}	0.40
		X _{max}	2.44
	mg/dL	n	188
Glu		Slope	1.00
		r	1.000
		intercept	1.17
		X_{min}	24
		X _{max}	671
Hct	%PCV	n	229
		Slope	1.02
		r	0.993
		intercept	-0.36
		X _{min} (%PCV)	18

Test	Units		Comparative Method i-STAT 1W
		X _{max} (%PCV)	70
pН		n	187
'		Slope	0.990
		r	0.999
		intercept	0.075
		X_{min}	6.592
		X _{max}	8.189
P O ₂	mmHg	n	192
		Slope	0.986
		r	0.998
		intercept	0.0
		X_{min}	9
		X _{max}	705
P CO ₂	mmHg	n	149
		Slope	0.989
		r	0.999
		intercept	0.3
		X_{min}	5.1
		X_{max}	129.8

FACTORS AFFECTING RESULTS

The following substances were evaluated in plasma for relevant analytes at the test concentrations recommended in CLSI guideline EP7-A2 ¹⁰ unless otherwise noted. For those identified as an interferant the interference is described.

Substance	Test Concentration (mmol/L)	Analyte	Interference (Yes/No)	Comment
Acetaldehyde	0.045 ¹¹	Glu	No	
		Na	No	
Acetaminophen	1.32	K	No	
Acetaililiopheil	1.32	iCa	Yes	Decreased results
		Glu	No	
Acetaminophen	0.132 11	iCa	No	
(therapeutic)	0.132	Glu	No	
	10.2	Na	No	
Acatulovataina		K	No	
Acetylcysteine		iCa	Yes	Decreased results
		Glu	Yes	Decreased results
Acetylcysteine	0.3 12 13	iCa	No	
(therapeutic)	0.3 0	Glu	No	
Acetoacetate	2.0	Glu	No	
		Na	No	
Accorbata	0.34	K	No	
Ascorbate	0.34	iCa	No	
		Glu	No	
Bromide	37.5	Na	Yes	Increased results. Use another method.

Substance	Test Concentration (mmol/L)	Analyte	Interference (Yes/No)	Comment
		К	Yes	Increased results and rate of star (***) outs. Use another method.
		iCa	Yes	Increased results. Use another method.
		Glu	Yes	Decreased results. Use another method.
		Hct	Yes	Increased rate of star (***) outs
		Na	No	
Bromide		K	No	
(therapeutic)	2.5 14 15 16	iCa	No	
(inerapedile)		Glu	Yes	Decreased results
		Hct	No	
		Na	No	
β-Hydroxybutyrate	6.0 ¹⁷	K	No	
p-riyaroxybatyrate	0.0	iCa	No	
		Glu	No	
Dopamine	0.006	Glu	No	
Formaldehyde	0.133 11	Glu	No	
Hydroxyurea	0.92	Glu	Yes	Increased results. Use another method.
	6.6	Na	No	
		K	No	
Lactate		iCa	Yes	Decreased results by up to 0.07 mmol/L.
		Glu	No	
Leflunomide	0.03	iCa	Yes	Decreased results
		Na	No	
Magnesium	1.0	K	No	
Chloride	1.0	iCa	Yes	Increased results by up to 0.04 mmol/L.
Maltose	13.3	Glu	No	
		Na	Yes	Increased results
Nithiodote (sodium	16.7 ¹⁸	K	Yes	Decreased results
thiosulfate)	10.7	iCa	Yes	Decreased results
		Glu	Yes	Decreased results
Pyruvate	0.31	Glu	No	
		Na	No	
Salicylate	4.34	K	No	
•	4.54	iCa	Yes	Decreased results
		Glu	No	
Salicylate (therapeutic)	0.5 ¹⁹	iCa	Yes	Decreased results by up to 0.03 mmol/L
Thiocyanate	6.9	iCa	Yes	Decreased results. Use another method
		Glu	Yes	Decreased results
Thiocyanate (therapeutic)	0.5 11	Glu	No	
Uric Acid	1.4	Glu	No	

The degree of interference at concentrations other than those reported above might not be predictable. It is possible that interfering substances other than those tested may be encountered.

- Relevant comments regarding interference of Acetaminophen, Acetylcysteine, Bromide, Hydroxyurea, Leflunomide, Nithiodote, and Salicylate are noted below:
 - Acetaminophen has been shown to interfere with i-STAT ionized calcium results at a concentration proscribed by the CLSI guideline, 1.32 mmol/L, which represents a toxic concentration. Acetaminophen at 0.132 mmol/L, which represents the upper end of the therapeutic concentration, has been shown not to significantly interfere with i-STAT ionized calcium results.
 - Acetylcysteine has been tested at two levels: the CLSI recommended level of 10.2 mmol/L and a
 concentration of 0.30 mmol/L. The latter is 3 times the peak therapeutic plasma concentration
 associated with treatment to reverse acetaminophen poisoning. APOC has not identified a
 therapeutic condition that would lead to levels consistent with the CLSI recommended level.
 - o Bromide has been tested at two levels: the CLSI recommended level and a therapeutic plasma concentration level of 2.5 mmol/L. The latter is the peak plasma concentration associated with halothane anesthesia, in which bromide is released. APOC has not identified a therapeutic condition that would lead to levels consistent with the CLSI recommended level.
 - O Hydroxyurea has been shown to interfere with glucose results at 0.92 mmol/L. Hydroxyurea is a DNA synthesis inhibitor used in the treatment of various forms of cancer, sickle cell anemia, and HIV infection. This drug is used to treat malignancies including melanoma, metastatic ovarian cancer, and chronic myelogenous leukemia. It is also used in the treatment of polycythemia vera, thrombocythemia, and psoriasis. At typical doses ranging from 500 mg to 2 g/day, concentrations of hydroxyurea in patients' blood may be sustained at approximately 100 to 500 μmol/L. Higher concentrations may be observed soon after dosing or at higher therapeutic doses.
 - Leflunomide has been shown to interfere with ionized calcium results at 0.03 mmol/L. Leflunomide is an isoxazole immunomodulatory agent that inhibits dihydroorotate dehydrogenase, an enzyme involved in *de novo* pyrimidine synthesis, and that has antiproliferative activity. It is used in the treatment of some immune diseases. Following oral administration, leflunomide is metabolized to an active metabolite, teriflunomide, which is responsible for essentially all its *in vivo* activity. The active metabolite teriflunomide reaches a plasma concentration of 8.5 μg/mL (0.031 mmol/L) after a 100 mg loading dose and the steady state concentration is maintained at 63 μg/mL (0.23 mmol/L) after 24 weeks of maintenance dose at 25 mg/day ²⁰ when treating inflammatory polyarthropathy.
 - Nithiodote (sodium thiosulfate) has been shown to interfere with sodium, potassium, ionized calcium and glucose results at 16.7 mmol/L. Nithiodote (sodium thiosulfate) is indicated for the treatment of acute cyanide poisoning. The journal article titled "Falsely increased chloride and missed anion gap elevation during treatment with sodium thiosulfate" indicated that sodium thiosulfate could be used in the treatment of calciphylaxis indicating that "the highest concentration likely to be seen in plasma [is] after infusion of a 12.5 g dose of sodium thiosulfate pentahydrate. Assuming that the 12.5 g dose of sodium thiosulfate pentahydrate is distributed in a typical blood volume of 5 L with a hematocrit of 40%, the peak sodium thiosulfate plasma concentration expected is 16.7 mmol/L." 18
 - Salicylate has been shown to significantly decrease ionized calcium results at a concentration proscribed by the CLSI guideline, 4.34 mmol/L, which represents a toxic concentration. Salicylate at 0.5 mmol/L, which represents the upper end of the therapeutic concentration, has been shown to decrease ionized calcium results by approximately 0.03 mmol/L.

OTHER FACTORS AFFECTING RESULTS

Factor	Analyte	Effect				
Sodium heparin	Na	Sodium heparin may increase sodium results up to 1 mmol/L. ²¹				
Venous stasis	iCa	Venous stasis (prolonged tourniquet application) and forearm exercise may increase ionized calcium due to a decrease in pH caused by localized production of lactic acid. ²²				
	рН	Venous stasis (prolonged tourniquet application) and forearm exercise may decrease pH due to localized production of lactic acid.				

Factor	Analyte	Effect
- 40101	Amanyto	Low hematocrit results can be caused by contamination of flush solutions
		in arterial or venous lines.
	11.4	Back flush a line with a sufficient amount of blood to remove intravenous
Line draw	Hct	solutions, heparin, or medications that may contaminate the sample.
		Five to six times the volume of the catheter, connectors, and needle is
		recommended.
		Heparin binds calcium. Each unit of heparin added per mL of blood will
		decrease ionized calcium by 0.01 mmol/L ²² Therefore, the correct ratio
		of heparin anticoagulant to blood must be achieved during sample
Heparin	iCa	collection. Intravenous injection of 10,000 units of heparin has been shown in adults to cause a significant decrease of ionized calcium of
		about 0.03 mmol/L. ²² Use only non-heparinized sample transfer
		devices when using the i-STAT system aqueous control and calibration
		verification materials.
	:0-	Exposing the sample to air will cause an increase in pH due to loss of
	iCa	CO ₂ , which will decrease ionized calcium.
		Exposure of the sample to air will cause an increase in PO2 when values
Exposing the	P O ₂	are below 150 mmHg and a decrease in PO2 when values are above
sample to air		150 mmHg (approximate P O ₂ of room air).
33p.3 13 3	pH	
	P CO ₂	Exposing the sample to air allows CO ₂ to escape which causes P CO ₂ to
	HCO₃	decrease and pH to increase and HCO ₃ and TCO ₂ to be underestimated.
	TCO ₂	Hemodilution of the plasma by more than 20% associated with priming
	Na	cardiopulmonary bypass pumps, plasma volume expansion or other
		fluid administration therapies using certain solutions may cause clinically
	iCa	significant error on sodium, chloride, ionized calcium and pH results.
Hemodilution		These errors are associated with solutions that do not match the ionic
		characteristics of plasma. To minimize these errors when hemodiluting
	pН	by more than 20%, use physiologically balanced multi-electrolyte
		solutions containing low-mobility anions (e.g., gluconate).
0.11	5 00	Do not ice samples before testing as PO ₂ results may be falsely elevated
Cold	P O ₂	in cold samples. Do not use a cold cartridge as PO ₂ results may be
temperature	K	falsely decreased if the cartridge is cold. Potassium values will increase in iced specimens
		If heparinized whole blood is allowed to stand before testing, potassium
	K	values will first decrease slightly, then increase over time.
		Glucose values will decrease in whole blood samples over time. Venous
	Glu	blood glucose is as much as 7 mg/dL less than capillary blood glucose
		as a result of tissue utilization. ²³
Allowing blood	-11	pH decreases on standing anaerobically at room temperature at a rate
to stand (without	рН	of 0.03 pH units per hour. ¹
exposure to air)	P O ₂	Standing anaerobically at room temperature will decrease PO2 at a rate
exposure to all)		of 2–6 mmHg per hour. ¹
	P CO ₂	Allowing blood to stand (without exposure to air) before testing will
		increase P CO ₂ by approximately 4 mmHg per hour.
	HCO ₃	Calculated HCO ₃ and TCO ₂ results are over-estimated, if blood is
	TCO ₂	allowed to stand (without exposure to air), due to metabolic processes.
		Serum Potassium results may be 0.1 to 0.7 mmol/L higher than Potassium results from anticoagulated samples due to the release of
Sample type	К	Potassium from platelets ² and red blood cells during the clotting
		process.
0	11.4	Samples from 1 mL syringes should not be used to determine hematocrit
Sample mixing	Hct	if testing is delayed.

Factor	Analyte	Effect		
Hemolysis			obtained from skin punctu	re samples may vary due to
	K	hemolysis or an increase in tissue fluid from improper technique during		
		the collection procedure.		
Under fill or partial draw	P CO ₂	The use of partial draw tubes (evacuated tubes that are adjusted to draw less than the tube volume, e.g., a 5 mL tube with enough vacuum to draw only 3 mL) is not recommended due to the potential for decreased PCO_2 , HCO ₃ and TCO ₂ values. Underfilling blood collection tubes may also cause decreased PCO_2 , HCO ₃ and TCO ₂ results. Care must be		
	НСО₃			
	TCO ₂	taken to eliminate "bubbling" of the sample with a pipette when filling a cartridge to avoid the loss of CO ₂ in the blood.		
pH dependence	Glu	The dependence of the i-STAT glucose test with respect to pH is as follows: values below pH 7.4 at 37 °C decrease results by approximately 0.9 mg/dL (0.05 mmol/L) per 0.1 pH unit. Values above pH 7.4 at 37 °C increase results by approximately 0.8 mg/dL (0.04 mmol/L) per 0.1 pH unit.		
P O ₂ dependence	Glu	The dependence of the i-STAT glucose test with respect to PO_2 is as follows: oxygen levels of less than 20 mmHg (2.66 kPa) at 37 °C may decrease results.		
Method of calculation	sO ₂	Calculated sO ₂ values from a measured P O ₂ and an assumed oxyhemoglobin dissociation curve may differ significantly from the direct measurement. ³		
Clinical conditions	HCO₃	Causes of primary metabolic acidosis (decrease calculated HCO ₃) are ketoacidosis, lactate acidosis (hypoxia), and diarrhea. Causes of primary metabolic alkalosis (increase calculated HCO ₃) are vomiting and antacid treatment.		
Erythrocyte sedimentation	Hct	 The measurement of certain blood samples with high erythrocyte sedimentation rates (ESR) may be affected by analyzer angle. While testing blood samples, beginning 90 seconds after the cartridge is inserted, the analyzer should remain level until a result is obtained. A level surface includes running the handheld in the downloader/ recharger. Hematocrit results can be affected by the settling of red blood cells in the collection device. The best way to avoid the effect of settling is to test the sample immediately. If there is a delay in testing of a minute or more, the sample must be remixed thoroughly. 		
White Blood Cell Count (WBC)	Hct	Grossly elevated white blood cell counts may increase results.		
Lipids	Hct	Abnormally high lipids may increase results. Interference from lipids will be about two thirds the size of the interference from protein.		
-		Hematocrit results	are affected by the level of	of total protein as follows:
Total Protein	Hct	Displayed Result	Total Protein (TP) < 6.5 g/dL	Total Protein (TP) > 8.0 g/dL
		HCT < 40% PCV	Hct decreased by ~1%	Hct increased by ~1% PCV
			PCV for each decrease of	for each increase of 1 g/dL
			1 g/dL TP	TP
		HCT > 40% PCV	Hct decreased by ~0.75% PCV for each decrease of 1 g/dL TP	Hct increased by ~0.75% PCV for each increase of 1 g/dL TP
		Total protein	levels may be low in r	neonatal and burn patient
		populations, as well as in additional clinical populations listed in Statland. ⁶ Total protein levels may also be decreased in patients undergoing cardiopulmonary bypass (CPB) or extracorporeal membrane oxygenation (ECMO) and with patients receiving large		

Factor	Analyte	Effect		
		 volumes of saline-based intravenous (IV) fluids. Care should be taken when using hematocrit results from patients with total protein levels below the adult reference range (6.5 to 8 g/dL). The CPB sample type can be used to correct the hematocrit result for the dilutional effect of the pump prime in cardiovascular surgery. The CPB algorithm assumes that cells and plasma are diluted equally and that the pump priming solution has no added albumin or other colloid or packed red blood cells. Since perfusion practices vary, it is recommended that each practice verify the use of the CPB sample type and the length of time in which the CPB sample type should be used during the recovery period. Note that for hematocrit values above 30% PCV, the CPB correction is ≤ 1.5% PCV; the size of the correction at this level should not impact transfusion decisions. 		
Sodium	Hct	The sample electrolyte concentration is used to correct the measured conductivity prior to reporting hematocrit results. Factors that affect sodium will therefore also affect hematocrit.		
Propofol (Diprivan®) or thiopental sodium	P CO ₂	The use of CG8+ cartridge is recommended, which is free from clinically significant interference at all relevant therapeutic doses.		
P O ₂ sensitivity	P CO ₂	In patient samples where the PO_2 is > 100 mmHg above the normal range (80-105 mmHg), an increase in PCO_2 of approximately 1.5 mmHg (with a range of 0.9 to 2.0 mmHg) may be observed for every 100 mmHg increase in PO_2 . For example, if an oxygenated patient has a measured PO_2 of 200 mmHg, and a normal PO_2 is 100 mmHg, the impact to the PCO_2 result may be increased by approximately 1.5 mmHg.		

KEY TO SYMBOLS

Symbol	Definition/Use
2 "	2 months room temperature storage at 18-30°C
	Use by or expiration date. An expiration date expressed as YYYY-MM-DD means the last day the product can be used.
LOT	Manufacturer's lot number or batch code. The lot number or batch will appear adjacent to this symbol.
Σ	Sufficient for <n> tests</n>
EC REP	Authorized representative for Regulatory Affairs in the European Community.
1	Temperature limitations. The upper and lower limits for storage are adjacent to upper and lower arms.
REF	Catalog number, list number, or reference
(2)	Do not reuse.
**	Manufacturer
$\bigcap_{\mathbf{i}}$	Consult instructions for use or see System Manual for instructions.
IVD	In vitro diagnostic medical device
(€	Compliance to the European directive on <i>in vitro</i> diagnostic devices (98/79/EC)
Rx ONLY	For prescription use only.

Additional Information: To obtain additional product information and technical support, refer to the company website at www.pointofcare.abbott.

References

- 1. Pruden EL, Siggard-Andersen O, Tietz NW. Blood Gases and pH. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.
- 2. Tietz NW, Pruden EL, Siggaard-Andersen O. Electrolytes. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.
- 3. CLSI. Blood Gas and pH Analysis and Related Measurements; Approved Guideline. *CLSI document C46-A*. 2001.
- 4. Evaluation of Formed Elements of Blood. In: Bower JD, Ackerman PG, Toto G, eds. *Clinical Laboratory Methods*. St. Louis: The C.V. Mosby Company; 1974.
- 5. Young DS. *Effects of Drugs on Clinical Laboratory Tests*. 3rd ed. ed. Washington, DC: American Association of Clinical Chemistry; 1990.
- 6. Statland BE. *Clinical Decision Levels for Lab Tests*. Oradell, NJ: Medical Economic Books; 1987.
- 7. Painter PC, Cope JY, Smith JL. Reference Ranges, Table 41–20. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.
- 8. CLSI. Procedure for Determining Packed Cell Volume by the Microhematocrit Method; Approved Standard-Third Edition. *CLSI document H07-A3*. 2000.
- 9. Clinical and Laboratory Standards Institute (CLSI). Measurement Procedure Comparison and Bias Estimation Using Patient Samples; Approved Guideline—Third Edition. *CLSI document EP09-A3*. 2013.
- 10. Clinical and Laboratory Standards Institute. Interference Testing in Clinical Chemistry; Approved Guideline—Second Edition. *CLSI document EP7-A2*. 2005.
- 11. Wu AHB. Tietz Clinical Guide to Laboratory Tests: Elsevier Health Sciences; 2006.
- 12. Whillier S, Raftos JE, Chapman B, Kuchel PW. Role of N-acetylcysteine and cystine in glutathione synthesis in human erythrocytes. *Redox Report*. 2009;14(3):115-121.
- 13. Ventura P, Panini R, Pasini MC, Scarpetta G, Salvioli G. N-acetyl-cysteine reduces homocysteine plasma levels after single intravenous administration by increasing thiols urinary excretion. *Pharmacological Research*. 1999;40(4):345-350.
- 14. Hankins DC, Kharasch ED. Determination of the halothane metabolites trifluoroacetic acid and bromide in plasma and urine by ion chromatography. *Journal of Chromatography B: Biomedical Applications*. May 1997;692(2):413-418.
- 15. Kharasch ED, Hankins D, Mautz D, Thummel KE. Identification of the enzyme responsible for oxidative halothane metabolism: Implications for prevention of halothane hepatitis. *Lancet*. May 1996;347(9012):1367-1371.
- 16. Morrison JE, Friesen RH. Elevated serum bromide concentrations following repeated halothane anaesthesia in a child. *Canadian Journal of Anaesthesia*. October 1990;37(7):801-803.

- 17. Charles RA, Bee YM, Eng PHK, Goh SY. Point-of-care blood ketone testing: Screening for diabetic ketoacidosis at the emergency department. *Singapore Medical Journal*. November 2007;48(11):986-989.
- 18. Wendroth SM, Heady TN, Haverstick DM, et al. Falsely increased chloride and missed anion gap elevation during treatment with sodium thiosulfate. *Clinica Chimica Acta*. April 2014;431:77-79.
- 19. Borthwick GM, Johnson AS, Partington M, Burn J, Wilson R, Arthur HM. Therapeutic levels of aspirin and salicylate directly inhibit a model of angiogenesis through a Cox-independent mechanism. *FASEB Journal*. October 2006;20(12):2009-2016.
- 20. Sanofi-Aventis Canada Inc. Product Monograph PrARAVA® Submission, Control No.: 187857. Date of Revision: December 23, 2015. Available at: http://products.sanofi.ca/en/arava.pdf.
- 21. Tips on Specimen Collection. In: Mark Zacharia, ed. *Vol 1. Monograph of Medical Laboratory Observer's "Tips from the Clinical Experts"*. Montvale NJ: Medical Economics in collaboration with Becton, Dickinson and Company; 1997.
- 22. Fraser D, Jones G, Kooh SW, Raddle I. Calcium and Phosphate Metabolism. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.
- 23. Young DS, Bermes EW. Influence of Site Collection on Blood Gases and pH. In: C.A. Burtis and E.R. Ashwood, ed. *Tietz Textbook of Clinical Chemistry*. Second Edition ed. Philadelphia: W.B. Saunders Company; 1994.

i-STAT is a trademark of the Abbott Group of companies.

Diprivan is a registered trademark of the AstraZeneca group of companies.







©2022 Abbott Point of Care Inc. All rights reserved. Printed in USA.

